

Industrial Processes and Product Use Sector

Regional African Workshops on REDD+ National Forest Monitoring Systems and Greenhouse Gas National Inventory Systems

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Outline

- Overview
- What gases?
- What sources?
- What features?
- Important for non-Annex I Parties?
- Methods for each category





Emissions from Industrial Processes

- Emissions from manufacturing processes that chemically or physically transform materials: e.g.,
 - CO₂ released from calcination of limestone (CaCO₃) in cement production
 - CO₂ generated from use of coke as a reducing agent in production of iron
 - HFC-23 generated as by-product from production of HCFC-22
- ➤ Emissions from fuel combustion in industrial activities are NOT included.



Emissions from Product Use

- ➤ GHGs are used in products and eventually released to the atmosphere: e.g.,
 - HFCs and PFCs: Used as substitutes for ozone depleting substances (e.g., refrigerants)
 - SF₆: Used in electrical equipment for electrical insulation and current interruption
 - N₂O: Used as anesthesia
 - NMVOCs: Used as solvents





Reporting Categories

1996 Guidelines + GPG2000



Sector 2: Industrial Processes

2A: Mineral Products

2B: Chemical Industry

2C: Metal Production

2D: Other Production

2E: Production of Halocarbons and SF₆

2F: Consumption of Halocarbons and SF₆

2G: Other

Sector 3: Solvent and Other Product Use

2006 Guidelines



Sector 2: Industrial Processes and Product Use

2A: Mineral Industry

2B: Chemical Industry

2C: Metal Industry

2D: Non-Energy Products from Fuels and Solvent Use

2E: Electronics Industry

2F: Product Uses as Substitutes for ODS

2G: Other Product Manufacture and Use

2H: Other





Improvement since 1996 Guidelines

- > Combined two sectors in the 1996 GLs
- ➤ Improved to explicitly include more manufacturing sectors and product uses identified as sources of GHGs, e.g.:
 - Production of lead, zinc, titanium dioxide,
 petrochemicals, liquid crystal display (LCD), etc.
 - SF₆ and PFCs uses in military applications, accelerators, etc.

Formerly these emissions should have been estimated and reported in "2G Other", etc.

- > New gases, actual emission estimates
 - Explained later.





Improvement since 1996 Guidelines

- Estimation of actual annual emissions instead of "potential emissions"
 - ✓ In the 1996 Guidelines and Good Practice Guidance for a few sources, the simplest methodology estimates a "potential emission" rather than the actual annual emission.
 - This "potential emission" assumes all the emissions from an activity occur in the current year, ignoring the fact they will occur over many years (e.g. methane emissions from waste in landfills occurs over decades as the decay processes take place).
 - ✓ In the 2006 Guidelines, simple default methods estimate emissions when they occur, thus removing the need for potential emissions.
 - ✓ The removal of potential emission estimates also allows the emission reductions of abatement techniques to be properly estimated.





Improvement since 1996 Guidelines

- Demarcation between Energy and IPPU
 - Clearer and practical guidance has been given.
- ➤ Non-Energy Uses of Fossil Fuels
 - Clearer guidance has been introduced on emissions from non-energy uses of fossil fuels (lubricants, paraffin waxes).
 - Emissions from primary use should be reported in IPPU.
 - Emissions from secondary fate should be reported in Energy (in case of combustion for heat/energy), or in Waste (in case of incineration at disposal sites).
 - A method has been introduced for checking the completeness of carbon dioxide emission estimates from the non-energy uses.

TABLE 1.2 TYPES OF USE AND EXAMPLES OF FUELS USED FOR NON-ENERGY APPLICATIONS

Type of use	Example of fuel types	Product/process	Chapter
Feedstock	natural gas, oils, coal	ammonia	3.2
	naphtha, natural gas, ethane, propane, butane, gas oil, fuel oils	methanol, olefins (ethylene, propylene), carbon black	3.9
Reductant	petroleum coke	carbides	3.6
	coal, petroleum coke	titanium dioxide	3.7
	metallurgical cokes, pulverised coal, natural gas	iron and steel (primary)	4.2
	metallurgical cokes	ferroalloys	4.3
	petroleum coke, pitch (anodes)	aluminium ¹	4.4
	metallurgical coke, coal	lead	4.6
	metallurgical coke, coal	zinc	4.7
Non-energy	lubricants	lubricating properties	5.2
product	paraffin waxes	misc. (e.g., candles, coating)	5.3
	bitumen (asphalt)	road paving and roofing	5.4
	white spirit2, some aromatics	as solvent (paint, dry cleaning)	5.5

¹ Also used in secondary steel production (in electric arc furnaces) (see Chapter 4.2).

^{2.} Also known as mineral turpentine, petroleum spirits, industrial spirit ('SBP').



Gases

- > A wide variety of gases
 - CO₂, CH₄, N₂O
 - HFCs, PFCs, SF₆
 - Other halogenated gases
 - Ozone/aerosol precursors (e.g., NMVOCs)
- → H₂O and Gases controlled by the Montreal Protocol (e.g., CFCs, HCFCs) are not included.
- ➤ Under the UNFCCC, non-Annex I Parties:
 - should report CO₂, CH₄ and N₂O
 - are encouraged to report HFCs, PFCs, SF₆ and precursors
- > New gases may emerge in the future.





"New" gases in 2006 Guidelines

- Sources Identified in 2006 Guidelines

nitrogen trifluoride (NF₃)

CF₃I, CH₂Br₂, CHCl₃

 C_4F_6 , C_5F_8 , $c-C_4F_8O$

CH₂Cl₂ CH₃Cl

 $C_3F_7C(O)C_2F_5$

trifluoromethyl sulphur pentafluoride (SF₅CF₃)

halogenated ethers (e.g. $C_4F_9OC_2H_5$,

CHF,OCF,OC,F,OCHF,, CHF,OCF,OCHF,)

By-product & fugitive emissions Halogenatec **GWP in AR4** Compounds **Production**

Magnesium

production

Electronics

Industries

	KER V
WMO	UNEP

"New" gases

"those for which either significant concentrations or large trends in concentrations have been observed or a clear potential for future emissions has been identified." For example:

Nitrogen Trifluoride (NF₃)

- Used in manufacture of LCD displays, photovoltaic cells
- Currently 0.04 % of the impact of current human-produced CO₂ emissions
- Current Concentration: 0.454 ppt
- Increasing at 11 % per year
- GWP (100 yr time horizon, 4AR) 17,200

Trifluoromethyl Sulphur Pentafluoride (SF₅CF₃)

- By-product of fluorinated gas production other sources unknown (related to SF₆)
- Current Concentration: 0.16-0.18 ppt
- Increasing at 6.3 % per year
- GWP (100 yr time horizon, 4AR) 17,700





Sources

(For details, see the slides at the end of this file.)

- > A wide variety of industries and products
 - Mineral industry
 - cement production, lime production, etc.
 - Chemical industry
 - ammonia production, nitric acid production, petrochemical production, fluorochemical production, etc.
 - Metal industry
 - iron and steel production, aluminium production, magnesium production, etc.
 - Non-energy products from fuels & solvent use
 - Lubricant use, paraffin wax use, solvent use, etc.



Sources

(For details, see the slides at the end of this file.)

- > A wide variety of industries and products
 - Electronics industry
 - semiconductor manufacturing, TFT flat panel display manufacturing, etc.
 - Product uses as ODS substitutes
 - refrigeration and air conditioning, foam blowing agents, fire protection, etc.
 - Other product manufacture and use
 - electrical equipment, medical applications, propellant for pressure and aerosol products, etc.
- New sources (new industries, new products) may emerge in the future.

2 Industrial Processes and Product Use (Note 1, 2)	CO ₂	CH4	N ₂ O	HFCs	PFCs	SF ₆	Other halo- genated Gases (Note 3)
2A Mineral Industry							
2A1: Cement Production	X	÷					
2A2: Lime Production	X	÷					
2A3: Glass Production	X	÷					
2A4: Other Process Uses of Carbonates			L _				
2A4a: Ceramics	X	÷					
2A4b: Other Uses of Soda Ash	X	÷					
2A4c: Non Metallurgical Magnesia Production	X	÷					
2A4d: Other	X	÷					
2A5: Other	X	÷	÷				
2B Chemical Industry							
2B1: Ammonia Production	X	÷	÷				
2B2: Nitric Acid Production	÷	÷	X				
2B3: Adipic Acid Production	÷	÷	X				
2B4: Caprolactam, Glyoxal and Glyoxylic Acid Production	÷	÷	X				
2B5: Carbide Production	X	X	÷				
2B6: Titanium Dioxide Production	X	÷	÷				
2B7: Soda Ash Production	X	÷	÷				
2B8: Petrochemical and Carbon Black Production							
2B8a: Methanol	X	X	÷				
2B8b: Ethylene	X	X	÷				
2B8c: Ethylene Dichloride and Vinyl Chloride Monomer	X	X	÷				
2B8d: Ethylene Oxide	X	X	÷				
2B8e: Acrylonitrile	X	X	÷				
2B8f: Carbon Black	X	X	÷				
2B9: Fluorochemical Production (Note 4)							
2B9a: By-product Emissions (Note 5)				X	X	X	X
2B9b: Fugitive Emissions (Note 3)				X	X	X	X
2B10: Other	÷	÷	÷	÷	÷	÷	÷



née

TABLE 1.1 INDUSTRIAL PROCESSES AND PRODUCT USE CATEGORIES AND THEIR POSSIBLE EMISSIONS

2 Industrial Processes and Product Use (Note 1, 2)	CO ₂	CH ₄	N ₂ O	HFCs	PFCs	SF ₆	Other halo- genated Gases (Nota3)
2C Metal Industry							
2C1: Iron and Steel Production	X	X	÷				
2C2: Ferroalloys Production	X	X	÷				
2C3: Aluminium Production	X	÷			X		
2C4: Magnesium Production (Note 6)	X			X	X	X	X
2C5: Lead Production	X						
2C6: Zinc Production	X						
2C7: Other	÷	*	*	÷	*	÷	÷
2D Non-Energy Products from Fuels and Solvent Use (Note 7)							
2D1: Lubricant Use	X						
2D2: Paraffin Wax Use	X	÷	÷				
2D3: Solvent Use (Note 8)							
2D4: Other (Note 9)	÷	÷	÷				
2E Electronics Industry							
2E1: Integrated Circuit or Semiconductor (Note 10)	±		÷	X	X	X	X
2E2: TFT Flat Panel Display (Note 10) 2E3: Photovoltaics (Note 10)				X	X	X	X
2E3: Photovoltaics (Note 10)				X	X	X	X
2E4: Heat Transfer Fluid (Note 11)							X
2E5: Other	÷	÷	÷	÷	÷	÷	÷



TABLE 1.1 (CONTINUED) INDUSTRIAL PROCESSES AND PRODUCT USE CATEGORIES AND THEIR POSSIBLE EMISSIONS

2 Industrial Processes and Product Use (Note 1, 2)	CO ₂	CH ₄	N₂O	HFCs	PFCs	SF ₆	Other halo- genated Gases (Nota3)
2F Product Uses as Substitutes for Ozone Depleting Substa	nces						
2F1: Refrigeration and Air Conditioning							
2F1a: Refrigeration and Stationary Air Conditioning	÷			X	X		÷
2F1b: Mobile Air Conditioning	÷			X	X		÷
2F2: Foam Blowing Agents	÷			X	*		÷
2F3: Fire Protection	÷			X	X		÷
2F4: Aerosols				X	X		÷
2F5: Solvents (Note 12)				X	X		÷
2F6: Other Applications	÷	÷	÷	X	X		÷
2G Other Product Manufacture and Use							
2G1: Electrical Equipment							
2G1a: Manufacture of Electrical Equipment (Note 13)					X	X	÷
2G1a: Manufacture of Electrical Equipment (Note 13) 2G1b: Use of Electrical Equipment (Note 13)					X	X	÷
2G1c: Disposal of Electrical Equipment (Note 13)					X	X	÷
2G2: SF ₆ and PFCs from Other Product Uses							
2G2a: Military Applications 2G2b: Accelerators (Note 14)					*	X	÷
2G2b: Accelerators (Note 14)					*	X	÷
2G2c: Other					X	X	*
2G3: N ₂ O from Product Uses							
2G3a: Medical Applications			X				
2G3b: Propellant for Pressure and Aerosol Products							
2G3c: Other			X				
2G4: Other		÷		±			÷
2H Other							
2H1: Pulp and Paper Industry (Note 15)	±	÷					
2H2: Food and Beverages Industry (Note 15)	±	*					
2H3: Other	*	÷	÷				

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Features

- > Diversity of sources and gases
 - Difficult to exhaustively include all sources & gases
 - Identify and include major sources & gases at least
- > Various opportunities for GHG abatement
 - Capture and abatement at plants
 (e.g., N₂O destruction at nitric acid production plants)
 - Recovery at the end of product's life and subject to either recycled or destroyed (e.g., HFCs in refrigerators)
- Care required in treating confidential data obtained from private sectors



Restricted data and confidentiality

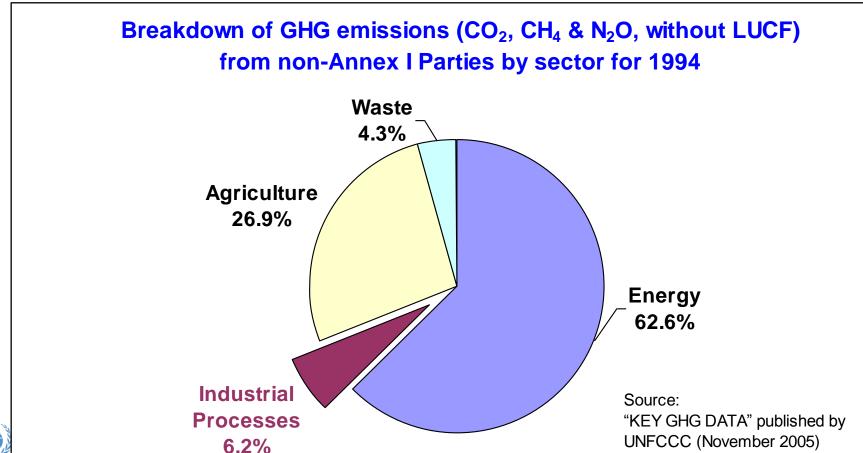
- Data providers might restrict access to information because it is confidential, unpublished, or not yet finalised.
- > Find solutions to overcome their concerns by:
 - ✓ explaining the intended use of the data,
 - ✓ agreeing, in writing, to the level at which it will be made public,
 - ✓ identifying the increased accuracy that can be gained through its use in inventories,
 - ✓ offering cooperation to derive a mutually acceptable data sets,
 - ✓ and/or giving credit/acknowledgement in the inventory to the data provided.





Importance for Non-Annex I Parties

➤ Tends to be considered less significant as compared to Energy and Agriculture







Importance for Non-Annex I Parties

- > Situation varies from country to country.
 - e.g., Peru reported 17.2% of GHGs was from Industrial Processes in 1994
- ➤ Significance of these sources may become greater in the future as development goes on.
- ➤ IP Sector emission estimation is important to find and make use of opportunities for GHG abatement.





How to estimate GHG emissions

➤ Typical Tier 1 – basic equation & default EF

Emission = $AD \times EF$

AD: Activity level data

(e.g., amount of material produced or consumed)

EF: Emission factor

(emission per unit of production or consumption)

- For some sources, a little more complex equation is used. See the IPCC Guidelines.
- ➤ Tier 2 (and 3): More detailed methods using:
 - Country-specific or plant-specific EFs
 - Direct measurement emission data
 - Data on GHG abatement / etc.





Chapter 2: Mineral Industry

Two pathways for CO₂ from carbonates

Calcination
 e.g. CaCO₃ + heat → CaO + CO₂

Acid-induced release

e.g.
$$CaCO_3 + H_2SO_4 \rightarrow CaSO_4 + H_2O + CO_2$$

Code	Category
2A1:	Cement Production
2A2:	Lime Production
2A3:	Glass Production
2A4:	Other Process Uses of Carbonates
2A4a:	Ceramics
2A4b:	Other Uses of Soda Ash
2A4c:	Non Metallurgical Magnesia Production
2A4d:	Other
2A5:	Other





Chapter 2: Mineral Industry

- Consistent approach based on carbonate content of inputs for all sources
- Inclusion of new guidance for other carbonates

TABLE 2.1 FORMULAE, FORMULA WEIGHTS, AND CARBON DIOXIDE CONTENTS OF COMMON CARBONATE SPECIES*					
Carbonate	Mineral Name(s)	Formula Weight	Emission Factor (tonnes CO ₂ /tonne carbonate)**		
CaCO ₃	Calcite*** or aragonite	100.0869	0.43971		
MgCO ₃	Magnesite	84.3139	0.52197		
CaMg(CO ₃) ₂	Dolomite***	184.4008	0.47732		
FeCO ₃	Siderite	115.8539	0.37987		
Ca(Fe,Mg,Mn)(CO ₃) ₂	Ankerite****	185.0225-215.6160	0.40822-0.47572		
MnCO ₃	Rhodochrosite	114.9470	0.38286		
Na ₂ CO ₃	Sodium carbonate or soda ash	106.0685	0.41492		

Source: CRC Handbook of Chemistry and Physics (2004)

^{****} Formulae weight range shown for ankerite assumes that Fe, Mg, and Mn are present in amounts of at least 1.0 percent.





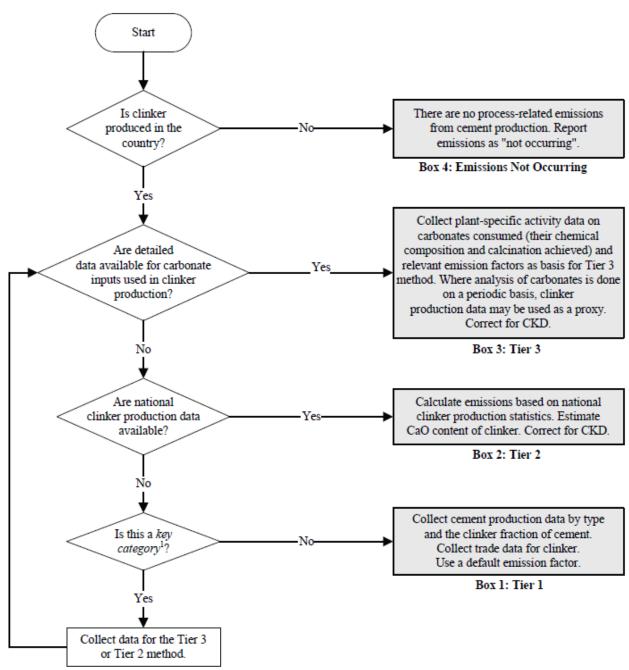
^{*} Final results (i.e., emission estimates) using these data should be rounded to no more than two significant figures.

^{**} The fraction of emitted CO₂ assuming 100 percent calcination; e.g., 1 tonne calcite, if fully calcined, would yield 0.43971 tonnes of CO₂.

^{***} Calcite is the principal mineral in limestone. Terms like high-magnesium or dolomitic limestones refer to a relatively small substitution of Mg for Ca in the general CaCO₃ formula commonly shown for limestone.

CO₂ from Cement Production

Figure 2.1 Decision tree for estimation of CO₂ emissions from cement production





CO₂ from Cement Production (Tier 1)

$$E_{CO2} = [\Sigma(M_{c,i} \times C_{cl,i}) - Im + Ex] \times EF_{clc}$$

 $E_{CO2} = CO_2$ emissions from cement production, tonnes

 $M_{c.i}$ = mass of cement produced of type *i*, tonnes

 $C_{cl,i}$ = clinker fraction of cement type *i*, fraction

Im = imports for consumption of clinker, tonnes

Ex = exports of clinker, tonnes

EF_{clc} = emission factor for clinker, tonnes CO₂/tonne clinker

- Default $EF_{clc} = 0.52$
 - Already corrected for cement kiln dust (CKD)
- "Activity data" is clinker production

$$= \left[\sum_{i} (M_{c,i} \times C_{cl,i}) - Im + Ex \right]$$





CO₂ from Cement Production (Tier 1)

- > To estimate clinker production:
 - National-level data should be collected on:
 - Cement production by type (Portland, masonry, etc.)
 - Clinker fraction by cement type
 - If detailed information on cement type is not available, multiply total cement production by:
 - Default Ccl = 0.75 (if blended/'masonry' is much)
 - Default Ccl = 0.95 (if all is essentially 'Portland')
 - Data should be obtained on the amount of clinker imported and exported.





CO₂ from Cement Production (Tier 2)

$E_{CO2} = M_{cl} \times EF_{cl} \times CF_{ckd}$

 $E_{CO2} = CO_2$ emissions from cement production, tonnes

 M_{cl} = mass of clinker produced, tonnes

EF_{cl} = emission factor for clinker, tonnes CO₂/tonne clinker

 CF_{cl} = emissions correction factor for CKD, dimensionless

 CKD not recycled to the kiln is considered to be 'lost' and associated emissions are not accounted for by the clinker.

EQUATION 2.5 CORRECTION FACTOR FOR CKD NOT RECYCLED TO THE KILN $CF_{ckd} = 1 + (M_d/M_{cl}) \bullet C_d \bullet F_d \bullet (EF_c/EF_{cl})$

Where:

CF_{ckd} = emissions correction factor for CKD, dimensionless

M_d = weight of CKD not recycled to the kiln, tonnes^a

Mcl = weight of clinker produced, tonnes

C_d = fraction of original carbonate in the CKD (i.e., before calcination), fraction^b

F_d = fraction calcination of the original carbonate in the CKD, fraction^b

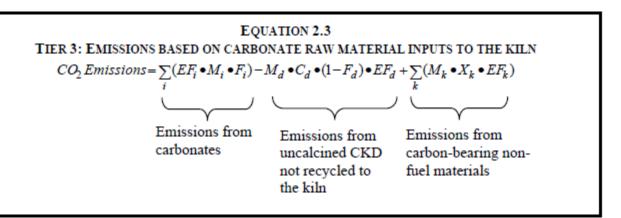
EF_c = emission factor for the carbonate (Table 2.1), tonnes CO₂/tonne carbonate

EF_{cl} = emission factor for clinker uncorrected for CKD (i.e., 0.51 tonnes CO₂/tonne clinker), tonnes CO₂/tonne clinker





CO₂ from Cement Production (Tier 3)



Where:

CO₂ Emissions = emissions of CO₂ from cement production, tonnes

EF_i = emission factor for the particular carbonate i, tonnes CO₂/tonne carbonate (see Table 2.1)

 M_i = weight or mass of carbonate i consumed in the kiln, tonnes

 F_i = fraction calcination achieved for carbonate i, fraction^a

M_d = weight or mass of CKD not recycled to the kiln (= 'lost' CKD), tonnes

C_d = weight fraction of original carbonate in the CKD not recycled to the kiln, fraction^b

F_d = fraction calcination achieved for CKD not recycled to kiln, fraction^a

EF_d = emission factor for the uncalcined carbonate in CKD not recycled to the kiln, tonnes CO₂/tonne carbonate^b

 M_k = weight or mass of organic or other carbon-bearing nonfuel raw material k, tonnes^c

 X_k = fraction of total organic or other carbon in specific nonfuel raw material k, fraction

EF_k = emission factor for kerogen (or other carbon)-bearing nonfuel raw material k, tonnes CO₂/tonne carbonate^c



Chapter 3: Chemical Industry

- Separation of CO₂ from urea use and production
- Various "new" sources added
- Soda Ash Production is included (formerly under 2A)
- Expanded method for HFC-23 and other F-gases by-product

Code	Category	Code	Category
2B1:	Ammonia Production	2B8b:	Ethylene
2B2:	Nitric Acid Production	2B8c:	Ethylene Dichloride and Vinyl Chloride Monomer
2B3:	Adipic Acid Production	2B8d:	Ethylene Oxide
2B4:	Caprolactam, Glyoxal and Glyoxylic Acid Production	2B8e:	Acrylonitrile
2B5:	Carbide Production	2B8f:	Carbon Black
2B6:	Titanium Dioxide Production	2B9:	Fluorochemical Production
2B7:	Soda Ash Production	2B9a:	By-product Emissions
2B8:	Petrochemical and Carbon Black Production	2B9b:	Fugitive Emissions
2B8a:	Methanol	2B10:	Other

Chapter 3: Chemical Industry

- CO₂ associated with urea production & use
 - Formerly, all these were implicitly included in CO₂ from Ammonia Production.
 - CO₂ recovered in the ammonia production process for urea production should be deducted from CO₂ emissions from 2B1 Ammonia Production.
 - CO₂ emissions from urea use/incineration should be reported in the category where they occur: e.g.,
 - Use of urea-based catalysts (Energy Sector)
 - Urea application to agricultural soils (AFOLU Sector)
 - Incineration of urea-based products (Waste Sector)
 - Thus, now, proper account can be taken for exports of urea produced in ammonia plants.

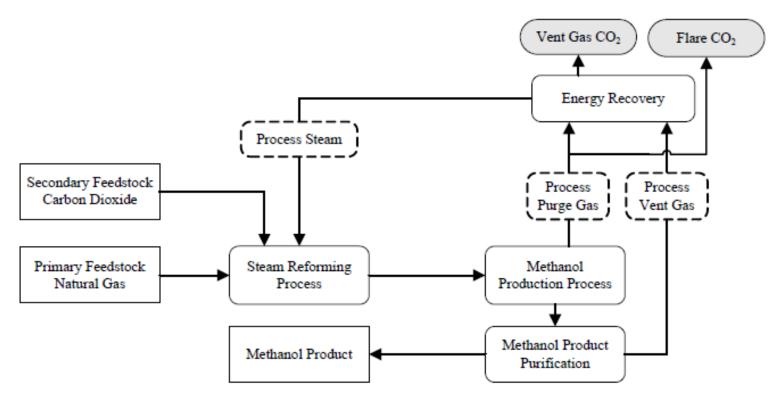




Emissions from petrochemical and carbon black production

e.g., methanol production

Figure 3.11 Methanol production feedstock-product flow diagram



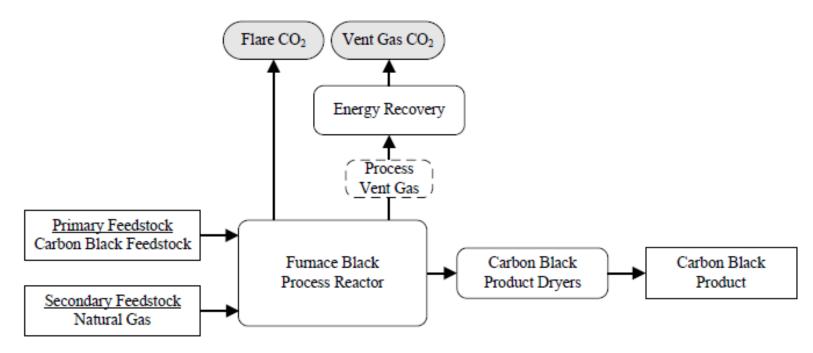




Emissions from petrochemical and carbon black production

e.g., carbon black production

Figure 3.15 Carbon black production feedstock-product flow diagram



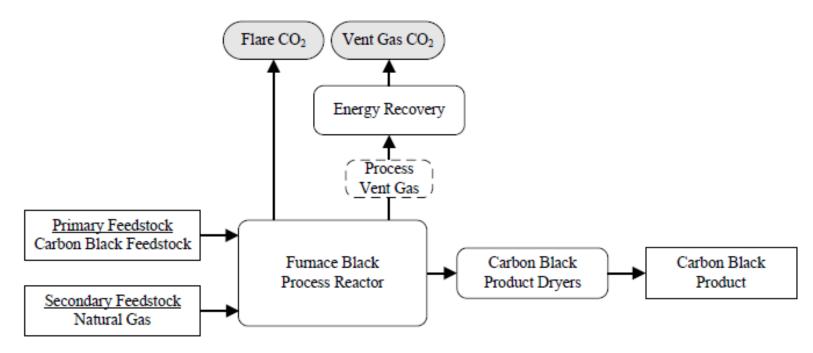




Emissions from petrochemical and carbon black production

e.g., carbon black production

Figure 3.15 Carbon black production feedstock-product flow diagram







Chapter 4: Metal Industry

- Detailed treatment of different production routes for iron and steel including DRI and metallurgical coke (Emissions from metallurgical coke should be reported under Energy Sector.)
- Improved guidance on production of ferroalloys
- New guidance on zinc and lead production

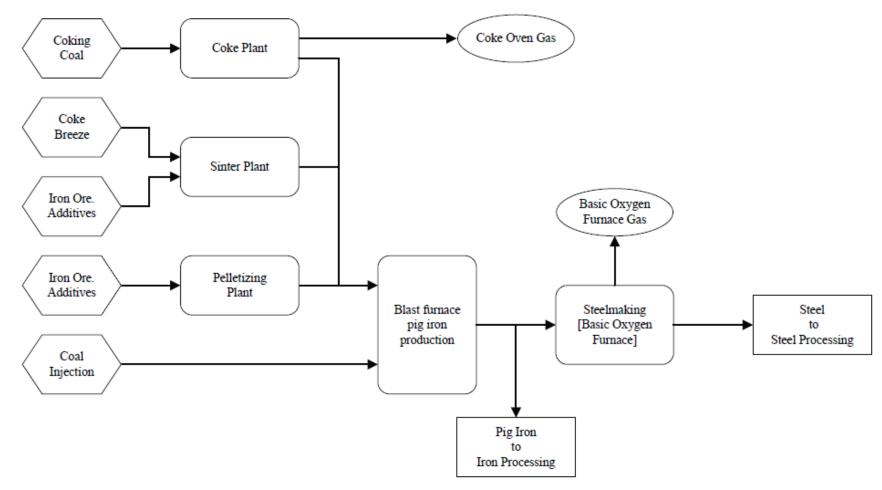
Code	Category	
2C1:	Iron and Steel Production	
2C2:	Ferroalloys Production	
2C3:	Aluminium Production	
2C4:	Magnesium Production	
2C5:	Lead Production	
2C6:	Zinc Production	
2C7:	Other	





CO₂ from Iron & Steel Production

Figure 4.1 Illustration of main processes for integrated iron and steel production*

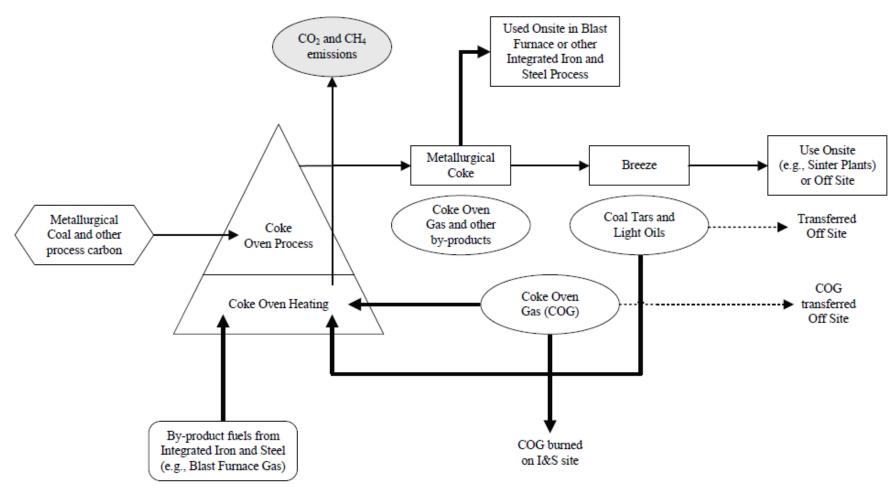


^{*} Modified from: European Conference on "The Sevilla Process: A Driver for Environmental Performance in Industry" Stuttgart, 6 and 7 April 2000, BREF on the Production of Iron and Steel conclusion on BAT, Dr. Harald Schoenberger, Regional State Governmental Office Freiburg, April 2000. (Schoenberger, 2000)



CO₂ from Iron & Steel Production

Figure 4.2 Illustration of coke production process (emissions reported in Category 1A of the Energy Sector)



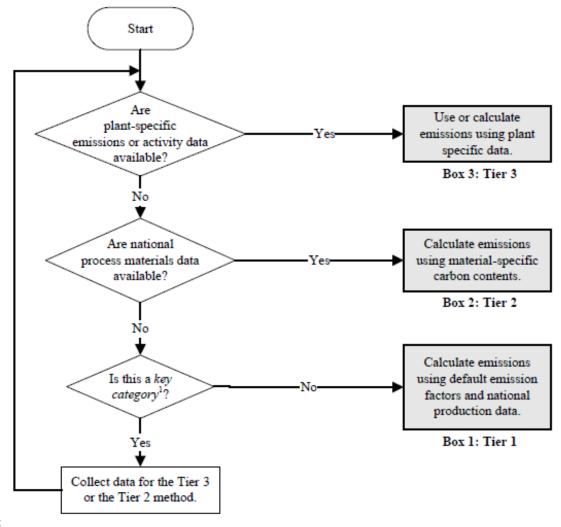
Note

Bold lines apply only to Onsite Coke Production at Integrated Iron and Steel Mill. Dashed lines apply to transfers of materials to 'Off Site processes.' 'Off Site processes' does not include Integrated Iron and Steel production processes, which are categorised as Onsite.

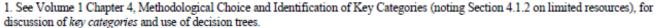


CO₂ from Iron & Steel Production

Figure 4.7 Decision tree for estimation of CO₂ emissions from iron and steel production









CO₂ from Iron and Steel Production (Tier 1)

$E_{CO2} = \Sigma(AD_i \times EF_i)$

 $E_{CO2} = CO_2$ emissions from iron & steel production, tonnes

 AD_i = quantity of material i, tonnes

 EF_i = emission factor for production of material i, tonnes CO_2 /tonne material i produced

Material *i* to be included:

- ➤ Crude steel from Basic Oxygen Furnace (BOF)
- ➤ Crude steel from Electric Arc Oxygen Furnace (EAF)
- ➤ Crude steel from Open Hearth Furnace (OHF)
- ▶Pig iron not converted to steel
- ➤ Direct reduced iron (DRI)
- **≻**Sinter
- **≻**Pellet



CO₂ from Iron and Steel Production (Tier 1)

> Emissions from metallurgical coke production should be reported in Energy Sector.

Default EFs are:

1.46 t-CO₂/t – BOF steel:

– EAF steel: $0.08 \text{ t-CO}_{2}/\text{t}$

1.72 t-CO₂/t – OHF steel:

– Pig iron: 1.35 t-CO₂/t

0.70 t-CO₂/t - DRI:

– Sinter: 0.20 t-CO₂/t

0.03 t-CO₂/t – Pellet:

Global average default $= 1.06 \text{ t-CO}_2/\text{t}$

> (If activity data on steel production for each process is not available, multiply total steel production by this EF.)





CO₂ from Iron and Steel Production (Tier 2)

EQUATION 4.9 CO₂ EMISSIONS FROM IRON & STEEL PRODUCTION (TIER 2)

$$\begin{split} E_{CO2,non-energy} = & \left[PC \bullet C_{PC} + \sum_{a} \left(COB_{a} \bullet C_{a} \right) + CI \bullet C_{CI} + L \bullet C_{L} + D \bullet C_{D} + CE \bullet C_{CE} \right. \\ & \left. + \sum_{b} \left(O_{b} \bullet C_{b} \right) + COG \bullet C_{COG} - S \bullet C_{S} - IP \bullet C_{IP} - BG \bullet C_{BG} \right] \bullet \frac{44}{12} \end{split}$$

Where, for iron and steel production:

 $E_{CO2, non-energy}$ = emissions of CO_2 to be reported in IPPU Sector, tonnes

PC = quantity of coke consumed in iron and steel production (not including sinter production), tonnes

 COB_a = quantity of onsite coke oven by-product a, consumed in blast furnace, tonnes

CI= quantity of coal directly injected into blast furnace, tonnes

L = quantity of limestone consumed in iron and steel production, tonnes

D = quantity of dolomite consumed in iron and steel production, tonnes

CE = quantity of carbon electrodes consumed in EAFs, tonnes

O_b = quantity of other carbonaceous and process material b, consumed in iron and steel production, such as sinter or waste plastic, tonnes

COG= quantity of coke oven gas consumed in blast furnace in iron and steel production, m³ (or other unit such as tonnes or GJ. Conversion of the unit should be consistent with Volume 2: Energy)

S = quantity of steel produced, tonnes

IP = quantity of iron production not converted to steel, tonnes

BG = quantity of blast furnace gas transferred offsite, m³ (or other unit such as tonnes or GJ. Conversion of the unit should be consistent with Volume 2: Energy)

 C_x = carbon content of material input or output x, tonnes C/(unit for material x) [e.g., tonnes C/tonne]



Chapter 5: Non-Energy Products from Fuels and Solvent Use

- Inclusion of previously separate sector on solvent use
- Consideration of use of fuels as lubricants, paraffin waxes, bitumen/asphalt and solvents
- Focuses on direct CO₂ emissions

Code	Category				
2D1:	Lubricant Use				
2D2:	Paraffin Wax Use				
2D3:	Solvent Use				
2D4:	Other				





Chapter 6: Electronics Industry

- Added guidance on production of PV cells, LCD and heat transfer fluids
- Inclusion of new gases applied in the industry
- Update of emission factors including treatment of abatement
- Inclusion of a new tier 1 method providing emission factors & activity data

Code	Category					
2E1:	Integrated Circuit or Semiconductor					
2E2:	TFT Flat Panel Display					
2E3:	Photovoltaics					
2E4:	Heat Transfer Fluid					
2E5: Other						





Chapter 7: Fluorinated Substitutes for ODS

 Tier 1 approach on "actual emissions" instead of "potential emissions"

Code	Category					
2F1:	Refrigeration and Air Conditioning					
2F1a:	Refrigeration and Stationary Air Conditioning					
2F1b:	Air Conditioning					
2F2:	Foam Blowing Agents					
2F3:	Fire Protection					
2F4:	Aerosols					
2F5:	Solvents					
2F6:	Other Applications					





Emissions of Fluorinated Substitutes for Ozone Depleting Substance (Refrigeration)

- > Sector includes
 - Commercial and Domestic Refrigeration
 - Commercial and Domestic Air Conditioning
 - Industrial Processes (chillers, cold storage, heat pumps etc.)
 - Vehicular Air Conditioning (cars, buses, trains)
- > Emissions occur from:
 - Leakage from equipment in use
 - Retirement scrapping of old equipment





Potential Emission & Actual Emission

➤ Two approaches to estimating emissions from F-gas consumption

✓ Potential emission estimates

 Assumes all the emissions from an activity occur in the current year, ignoring the fact they will occur over many years TABLE A2.1

CALCULATION SCHEME FOR POTENTIAL EMISSIONS
ACCORDING TO THE FORMER TIER 1a

REPORTING YEAR:

TYPE OF CHEMICAL (e.g., HFC-134a):

Production of chemical

- + Import of chemical in bulk
- Export of chemical in bulk
- Destruction of chemical
- Sum (potential emission of chemical)

Actual emission estimates

Estimates emissions that actually occur in the current year.

1996 Guidelines Tier 1 = Potential Emission

Tier 2 = Actual Emission

2006 Guidelines Tier 1, Tier 2 = Actual Emission

Potential Emission approach is used only for verification.





Better!!



Actual Emission Estimates

- ➤ Why are actual emission estimates better than potential emission estimates...? That is because:
 - ✓ It takes into account the time lag between consumption and emission, which may be considerable in some application areas, e.g., refrigeration.
 - ✓ Time lag results from the fact that a chemical is placed in new products and then slowly leaks out over many years.

Potential emission estimates do not take this fact into account at all!!







Total amount of substances contained in existing equipment, chemical stockpiles, foams and other products not yet released to the atmosphere

Imports of Gas

Imports of Gas in **Equipment**

Bank of Gas in **Equipment**

Manufacture of Gas





Difficulties

➤ However, estimation of actual emissions is not as easy as potential emissions, because it has to take the "bank" into account, which requires:

✓ Complex calculation as compared to very simple equation for potential emission estimates

$$Emissions_t = Bank_t \bullet EF + RRL_t$$

and

overcome by the software.

This can be

$$Bank_{t} = \sum_{i=t_{0}}^{t} \left(Production_{i} + Imports_{i} - Exports_{i} - Destruction_{i} - Emissions_{i-1} \right) - RRL_{t}$$

✓ Historic data on production, exports, imports, etc of chemicals (cf., potential emission estimates require only the current year data)



What about this difficulty...? Can we overcome it, too?



Yes, we can!!

- ➤ New IPCC software enables you to estimate actual emissions even if you do not have historic data if you have at least the data/information on:
 - Year of introduction of agent
 - Domestic production of agent in current year
 - Imports of agent in current year
 - Exports of agent in current year
 - Growth rate of sales of equipment that uses the agent
 - ✓ For example, in the case the data are available only for 2005 and 2010 while you know the chemical has been used since 1995...

(tonne)	1995	1996	1997	1998	1999	2000	2001	2002	2003	2004	2005	2006	2007	2008	2009	2010
Produced Quantity											26091					27925
Exported Quantity											18046					23963
Imported Quantity											9287					17222

Empirical assumption...

➤ In the absence of historic data on production, imports and exports, the calculations assume that the total market for equipment grows exponentially while the share of the market which is taken by the F-gas grows linearly between the year of introduction and the current year.

$$M(t+1) = (1+r) \times M(t) \times (t - t_0 + 2)/(t - t_0 + 1)$$

M(t) = Production, Exports or Imports of F-gas in year t

r = Growth rate of sales of equipment that uses the agent (fraction)

 t_0 = Year of introduction of agent





Chapter 8: Other Product Manufacture and Use

- SF₆ (and PFCs) from electrical equipment:
 - Replacement of three parallel Tier 3 mass balance methods by one flexible method
 - New tier 1 emission factors for regions and technologies
- Other sources: e.g. nuclear fuel cycle, military applications

	Code	Category	Code	Category
	2G1:	Electrical Equipment	2G2c:	Other
	2G1a:	Manufacture	2G3:	N ₂ O from Product Uses
	2G1b:	Use	2G3a:	Medical Applications
	2G1c:	Disposal	2G3b:	Propellant for Pressure and Aerosol Products
1E	2G2:	SF ₆ and PFCs from Other Product Uses	2G3c:	Other
	2G2a:	Military Applications	2G4:	Other
	2C2h.	A acalamatama		





Any questions?

Please note:

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